## Exercise 10.1 Line-Bond Notation

1. Each of the following molecules is drawn using line-bond notation. Draw the complete structural formula for each molecule. Show all bonds and atoms, including lone pairs; bond angles should be drawn accurately.
(a)

(b)

(c)

(d)

(e)

(f)

2. The complete structural formula for each of the following molecules has been drawn. Redraw each molecule using line-bond notation.
(a)

(b)

3. The condensed formula for each of the following molecules has been written. Draw each molecule using line-bond notation.
(a) $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{NHCH}_{3}$
(b) $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{OCH}_{2} \mathrm{CH}_{3}$
(c) $\mathrm{CF}_{3} \mathrm{CO}_{2} \mathrm{H}$
