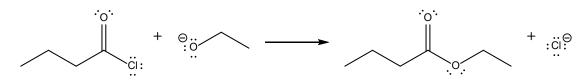
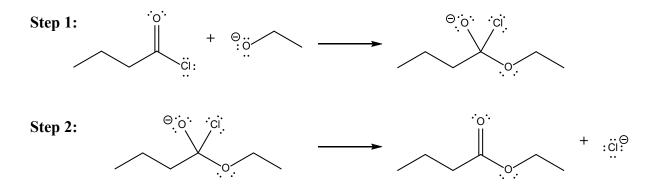
Answers to Exercise 11.3 Categories of Organic Reactions

1. At first glance the following reaction appears to be a <u>substitution</u>:

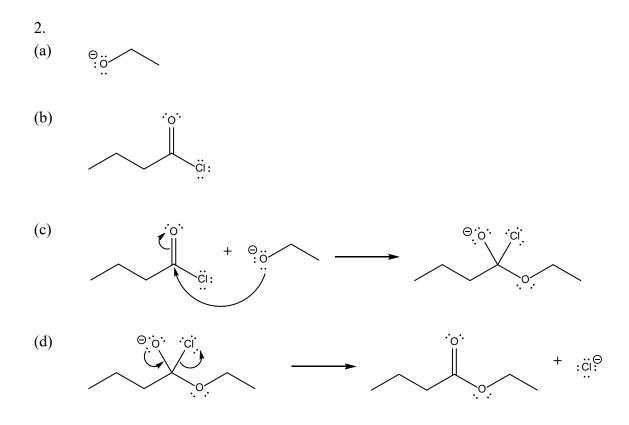


If you look at its mechanism, however, you can see that this is actually a 2-step process:



The first step is an <u>addition</u> while the second step is an <u>elimination</u>. As such, this reaction is more properly described as an <u>addition-elimination</u> reaction *(in which the first word refers to the classification of the first step and the second word refers to the classification of the second step)*.

Please note that you are not expected to be able to predict what the steps look like within a reaction class you have not previously seen. This question is only fair because the steps are shown.



Please note that arrows creating a new bond point directly to the atom to which the bond is being made; however, arrows increasing bond order (taking a single bond to a double bond, for example) point directly at the bond. Arrows showing the movement of an electron pair onto a single atom as a lone pair point directly at that atom. Arrows always start at the electron pair that is moving (either a lone pair or a bonding pair of electrons).