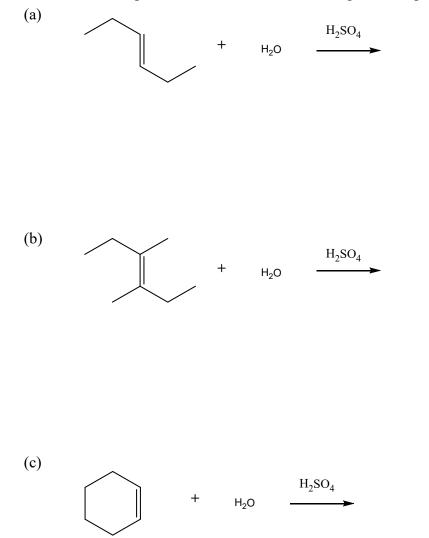
Exercise 11.5 Addition of H₂O to Alkenes

1. For each of the addition reactions below:

- draw curved arrows showing electron movement for the step in which the alkene is protonated,
- draw the carbocation intermediate formed and classify it as 1°, 2° or 3°,
- draw curved arrows showing electron movement when the carbocation reacts with water, and draw the product of this step of the reaction,
- draw curved arrows showing electron movement when the second intermediate is deprotonated, and draw the resulting addition product.



- 2. For each of the addition reactions below:
 - draw curved arrows showing electron movement for the step in which the alkene is protonated,
 - draw both possible carbocation intermediates and classify each as 1°, 2° or 3°,
 - identify the more stable carbocation intermediate; if it is resonance-stabilized, draw any other relevant resonance structures,
 - draw curved arrows showing electron movement when the more stable carbocation reacts with the water, and draw the product(s) of this step of the reaction,
 - draw curved arrows showing electron movement for deprotonation of the second intermediate(s), and draw the resulting addition product(s).

